

Chemistry 1A, General Chemistry

De Anza College Winter 2024 (1/7 - 3/29/2024)

Class times:

This is a class that meets every Tuesdays and Thursdays for lecture and twice per week for laboratory

section /CRN	Lecture Room S35	Laboratory Room SC2202
CHEM D001A25 / 32199	Tuesdays & Thursdays 4:30-5:45 (Leclerc)	Monday & Wednesday 11:30-2:20 (Wang)
CHEM D001A26 / 37908		Tuesdays & Thursdays 11:30-2:20 (Leclerc)

Instructors:

Margarete Leclerc, email: leclercmargarete@fhda.edu. **Office Hours:** Tuesday/Thursday: 3:20-4:20 in SC1200

Rose Wang, email: wangxiao@fhda.edu

Course Description:

This course provides an introduction to the structure and reactivity of matter at the molecular level, as well as an application of critical reasoning to modern chemical theory and structured numerical problem-solving. Students will learn the development of molecular structure from rudimentary quantum mechanics, including an introduction to ionic and covalent bonding; chemical problem solving involving both formula and reaction stoichiometry employing the unit analysis method, and be introduced to thermochemistry and a discussion of the first law of thermodynamics.

This course is divided into two separate instructional periods, the lecture and laboratory sections. The lecture portion is primarily devoted to the material discussion while the laboratory portion gives a chance for students to practice chemical experimentation. One registration code will enroll for the lecture and lab sections. Lecture and lab sections must be taken together to pass Chem 1A and will both go towards a single grade.

Prerequisites:

CHEM 25 or CHEM 30A or satisfactory score on the Chemistry Placement Test; MATH 114 or MATH 130 or equivalent

Advisory: EWRT 1A or EWRT 1AH or (EWRT 1AS and EWRT 1AT) or ESL 5

Required Course Material:

- Lecture Text:** Primary Lecture Text: OPEN STAX Chemistry, Second Edition. ISBN 978-1-947172-62-3. This is a free, online textbook and we will follow chapters 1-8 in this course.
- Access to Canvas:** Assignments and some course material will be posted on Canvas and you will need to upload completed assignments to the canvas site. You need a working camera connected to the internet and a way to create pdf files. Common cell phones are usually suitable. Please let me know if this is an issue for you.

3. **Aktiv Chemistry subscription for homework and quizzes.** We will use AktivChemistry as our online homework and in-class practice problem and lecture quiz platform this quarter. The homework and the in-class assignments will be graded and contribute to your final grade. You will need to sign up for this subscription through Canvas so that your Aktiv Chemistry account will be linked with our Canvas site. After signing up, you will have complimentary access for the first two weeks of the quarter. After this period, you will need to activate your paid subscription. (\$35)
4. **Supplementary Lecture text:** CHEMISTRY: The Molecular Nature of Matter and Change, Silberberg and Amateis, 9e. Any edition. This is a great textbook as well and could be used as a great reference. Practice examples from this book can be done for additional practice.
5. **Lab Notebook:** To record your experimental work and experimental analysis you will need a permanently bound, about 9"×11" notebook. Make sure it is not spiral bound but permanently bound, so no pages can be easily detached. Here are two examples: [Computation Notebook](#) or [Interactive Notetaking Composition Book](#).
6. **Lab Manual:** A printed copy of the lab manual: <https://www.deanza.edu/chemistry/Chem1A.html> Lab manuals must be read before performing each lab. Further instructions in lab section syllabi.
7. **Lab Safety Goggles:** You will need full safety goggles that seal on the sides, not just safety glasses. The goggles need to meet the ANSI Z87.1 or Z87+ specification.
8. **Scientific Calculator.** Logarithm and exponential functions required, No graphing calculators. You are encouraged to bring your calculator each day to work through examples as they are presented. Phones will not be allowed for calculations during tests so be sure to bring a calculator those days.

Registration details:

Class Registration. This class is a lecture and laboratory-based course, so the registration limit is strictly set at 30 students per section based on the number of people able to safely conduct experiments in the space provided.

Dropping the Course. Students that choose to drop this course are responsible for requesting a withdrawal through the admissions and records department **before** the deadline. Students who drop the class are to be also required to officially check-out of the lab locker. Failure to check out by the scheduled check-out date will result in fees and a block placed on future registrations.

Resources: Learn about [Student services](#) Academic support and Information about tutoring can be found at the [Math Science and Technology Resource Center](#) .

Academic Integrity: By enrolling in classes at De Anza College, you agree to the academic integrity policy and are held to all standards. Specifics can be found at [Academic Integrity \(deanza.edu\)](#) and it is your responsibility to understand what academic dishonesty involves. Cheating during an exam or quiz will not be tolerated and will result in zero for that quiz/exam regardless of what percentage of the work is from cheating. Worse than a 0 on an exam, I am required to report such incidents to the disciplinary committee, who will make a note of the incident on your transcript, which then becomes visible to 4 year colleges upon reviewing your transfer application.

Disability Service Support: De Anza is committed to providing support for students with disabilities. Please contact me as soon as possible if you require special accommodation and I will be happy to do what I can to help. For more information, visit [Disability Service Support](#)

Missing class: If you miss class on the first day of class, you will be dropped from the course unless previous arrangements have been made with the instructor. Regular participation in class is essential for success in class. **Please know that instructors may drop any student who does not participate in the first lab classes. Lab class is in-person and mandatory for this course. If you miss more than 3 lab periods or you do not complete more than two labs will result in a failing grade of F for the course.**

Grades/Evaluations:

Your grade will be determined through assignments and assessments in lecture and lab. Here are the approximate percentages that each assignment groups contributes to the final grade:

Assignment groups	Overall%
Aktiv Homework (10) weekly	10
Aktiv in-class assignments and 3 lecture quizzes	8
Lecture Exams (3)	34
Lecture Final (1)	14
Lecture Total	66
Lab Assignments Total (see specifics in lab syllabi)	34
Course total	100

Grade Assignment. Grade cut offs are as follows:

A+ (97), A (93), A- (90), B+ (87), B (83), B- (80), C+ (76), C (70), D (60), F (<60)

IMPORTANT NOTE: You must receive at least 55% on the lab assignments **and** homework **and** lecture exams for a passing grade in addition to achieving 70% or higher overall.

If you stop participating in the course after the “W” deadline, I may assign you an “FW” grade. Participation is determined by submitting assignments on a regular basis, so not submitting assignments on a regular basis for more than 2 weeks or not responding to email notifications for more than 2 weeks will be considered non-participation.

Assignments fall into the following categories.

Homework through Aktiv. Weekly practice is essential to master chemistry concepts. Before attempting the Aktiv Chemistry homework you need to have practiced solving questions from lecture and from the textbook. The homework on the Chem 101 platform is graded homework and you should use the homework to assess your learning. You will have 2 attempts at the homework. However, make sure that you can solve most homework problems (at least 70%!) without any assistance from notes, the internet, a friend, etc. for success in Midterms and final. If you have difficulty completing the homework questions without assistance, you need to seek out support and/or practice more examples. Completing the homework is strongly recommended and represents the minimum needed to practice the topics. You are strongly encouraged to go beyond the assigned problems and try others throughout the textbook or the supplementary texts. The lowest of 11 weekly homework assignments will be dropped.

In class assignments and lecture quizzes: There will be regular AktivChemistry assignments during class or other assignments throughout the course. Three quizzes will be given between the exams to make sure everyone is keeping up with the material throughout the quarter. The quizzes are worth 15 points each, will take about 15-20 minutes, and will be given at the beginning of class so late attendance may result in missing time for

the quiz. The quizzes help you to keep up throughout the course and will alert you if you need to study specific topics again before the midterm exams, I will drop the lowest 1 of the 3 lecture quizzes as well as 1 of the other lecture assignments.

Lecture Exams. There will be three (3) written (on paper) lecture exams to test comprehension throughout the quarter, the dates are indicated in the lecture schedule below. Exams will cover material from lectures, homework, and book chapters. If you have difficulties to complete the homework without outside help, you need to get additional support before you take the exam. Each midterm exam is worth **100 points** and the dates are given in the schedule below. There are no automatic make-up exams. Missing a midterm will result in zero credit without written proof for an excused absence such as a police report, an official doctors note, etc.

Lecture Final. A comprehensive final will be worth **125** pts and will cover all material from the course. The time is set by the final schedule for Thursday 3/28 4:00-6:00pm. You must take the final at this specific time to pass this class, so please do not sign up for this class if you can't make the final time.

Lab assignments. They consist of prelabs, lab quizzes, Lab experimentation which includes data acquisition and calculations, and written conclusions. One (1) Lab final will be given on the last day of the lab. More descriptive details are given in the lab syllabi.. The lab points represent 34% of your final grade.

Please note that I will NOT provide extra credit work at the end of the semester for students. Please work on the given assignments throughout the course. I will grant up to two specific extensions without late penalty for incomplete assignments, see also late work policy below.

Class policies, Participation & Attendance & Late Work Policy

Regular lecture participation is highly encouraged. There are no make-ups for missed lecture quizzes or in-class assignments. There are no automatic make-up dates for lecture exams. Please contact me if you have a documented medical or other emergency to address missed exams as soon as possible. You must take the final at the date indicated in the schedules in order to pass the class.

Participation, cooperation, respect and courtesy are expected. No food or beverages except water are to be consumed in the classroom. Cell phones must be turned to silent during class sessions. No cameras or recording devices of any kind allowed in this class at any time. I do post the slides we work through in lecture ahead of time.

A note on disruptive behavior: Professional behavior, conduct, communication (verbal and nonverbal), and language is expected at all times. The college will enforce all policies and procedures set forth in the *Standards of Student Conduct*:

(<https://go.boarddocs.com/ca/fhda/Board.nsf/goto?open&id=9U2UC77B2DA5>Links to an external site.). Any student disrupting the class may be asked to leave that class.

Lab class is in-person and mandatory for this course. Missing three labs will result in an automatic failing grade in the course. Please review the lab description below carefully and approach your lab instructor with any questions you may have in regard to lab attendance.

Late work is accepted within the grace period indicated on Canvas and Aktiv Chemistry. There is a

late submission penalty of 5% each day. The grace period for homework is 5 days and/or right before the midterm exam. Each student can get the late penalty waved or can request a specific extension twice per quarter.

Class Lecture

This class (Chem 1A) will cover chapters 1-8 from the assigned textbook. The lecture will serve to cover the most important aspects of the chapter. The quantitative aspects (aka calculations) of concepts will be exemplified in lecture as well. However, students are still responsible **to review the covered concepts and additional examples with the help of the textbook**. More details will be given in the respective Chapter modules on Canvas. Below are helpful tips that make learning easier.

1. Prepare for lecture by reading and previewing the textbook chapter before attending lecture.

This will make the presented material much easier to understand and you will be able to engage in exercises and discussion about the material. Pre-reading the textbook before class will allow you to sort the presented information more effectively and therefore will help retain the concepts.

2. Attend lecture. Attending lecture will clarify material and will also include additional active learning activities that will help you make deep connections with the material. In lecture additional information may be presented that is not in your textbook. Also, Chemistry concepts are built on previous concepts and foundational knowledge. If you miss too many lecture classes, you will increase the likelihood that you may fail the class. I will make the slides available through canvas.

3. Review the lecture material and complete practice problems in each sub-chapter we covered as you review material. By engaging with the material through problem solving, you actively learn the material! There is not enough time to go over every concept in detail in lecture, so re-reading the textbook in connection with problem solving is essential to master the concepts. Don't wait until the midterm exam is approaching: Review the material promptly such as within 1 day of the lecture!

4. Each week work out the chapter problems. Plan on spending at least 2-3 hours studying outside of class for each lecture hour (including lab lectures). That's at least 7-10 hours weekly! Extensive practice is the best way to ensure mastering the chem1A material. It is essential to practice the material promptly, so do the Chapter problems **in the same week as the material is covered**. There are plenty of additional problems in the textbook throughout the chapters as well as at the end of each chapter, which will prepare you for midterms and final.

5. Complete the HW as your self-assessment, treat it as a quiz/exam question. The homework should inform you if you have mastered the concepts of the chapter and uncover gaps you need to work on closing. Ultimately, you will need to be able to solve the questions without assistance from your notes, the textbooks, friends, the internet etc. to score high on the exams.

6. Don't fall behind. Make sure to set aside time to complete your assignments weekly by the due date. Cramming before exams without studying the material during the weeks leading up to an exam does not usually work. Also, in chemistry, each new topic will build on the previous, so it is essential to understand the topics as they are presented (hence do the practice problems). Following a lecture when you do not understand the previous material is not an effective method for learning and will lead to further problems. To avoid falling behind...

7. Get help. If you have a difficult time with a topic, it is your responsibility to get help promptly. There are plenty of resources for aiding in material comprehension, but it all starts with you making an effort to get this help. I strongly encourage you to find a study group, working with peers is extremely helpful for mastering material. Also, come to office hours to get any follow up questions answered.

Lecture schedule:

Tentative schedule: Here is the approximate timing for when we cover what material from Chemistry2E. This is subject to change. Your Canvas page will contain the most updated scheduling. All dates, including exams are subject to change throughout the quarter. The final exam date will not change.

WEEK OF	WEEK	TUESDAY	THURSDAY
1/8/2023	1	Chapter 1	Chapter 1/2
1/15/2023	2	Chapter 2	Chapter 3+ Quiz 1
1/22/2023	3	Chapter 3	Chapter 4
1/29/2023	4	Exam 1	Chapter 4
2/5/2023	5	Chapter 6	Chapter 6
2/12/2023	6	Chapter 7	Chapter 7 + Quiz 2
2/19/2023	7	Chapter 5	Exam 2
2/26/2023	8	Chapter 5	Chapter 5
3/5/2023	9	Chapter 8	Chapter 8 + Quiz 3
3/12/2023	10	Chapter 8	Exam 3
3/19/2023	11	Review	Review
3/26/2023	12	No class	4:00-6:00 pm

Student Learning Outcome(s):

- Identify and explain trends in the periodic table.
- Construct balanced reaction equations and illustrate principles of stoichiometry.
- Apply the first law of thermodynamics to chemical reactions.

Office Hours:

T,TH 03:20 PM 04:20 PM In-Person SC1200